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1 Chapter 11 Solution Thermodynamics: Theory In the
chemical, petroleum, and pharmaceutical industries
multicomponent gases or liquids commonly undergo
composition changes as the result of mixing and separation
processes, the transfer of species from one phase to another,
or chemical reaction.

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Convenience. here suggests the definition of a fundamental.
new property called the chemical potential, upon.

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What is the difference between kg-mass and kg force?

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Chapter 11 Thermodynamics 11.1 Heat, work and energy -

Heat is the energy that is the process of being transferred from one to another due to a temperature difference - Work is energy that is being transferred due to the action of a force -

James Joule's careful experiments led to the law of conservation of energy - Normally when a substance is heated, its particles gain kinetic energy - During a phase change, the heat energy increases the distance between particles.

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Thermodynamics ...

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Thermodynamics [Thermodynamics] is the sixth chapter in
the NCERT class 11 chemistry textbook. Thermodynamics is
a branch of science that deals with the relationship between
heat and other forms of energy. A part of the universe where
observations are made is called system.

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property of such systems depend strongly on composition as

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well as on temperature and pressure, our purpose in this chapter is to develop the theoretical foundation for application of thermodynamics to gas mixture and liquid solution. 3
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