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covalent bonds. The majority of covalent bonds form between nonmetallic elements. 8. Describe how the octet rule applies to covalent bonds using Lewis structures.

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Chapter 8 Covalent Bonding Pt III

242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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Chapter 8 Covalent Bonding and Molecular Structure 8-3 There are two types of repulsive forces between the two atoms result from the

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Chapter 6 Section 6 2 Covalent Bonding Answer Key section 1 answers chapter 8 covalent bonding and molecular structure 8 2 81 interactions between particles coulombs law matter is made up of atoms and ions that experience both attractive and repulsive forces the strength of the force holding oppositely

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